

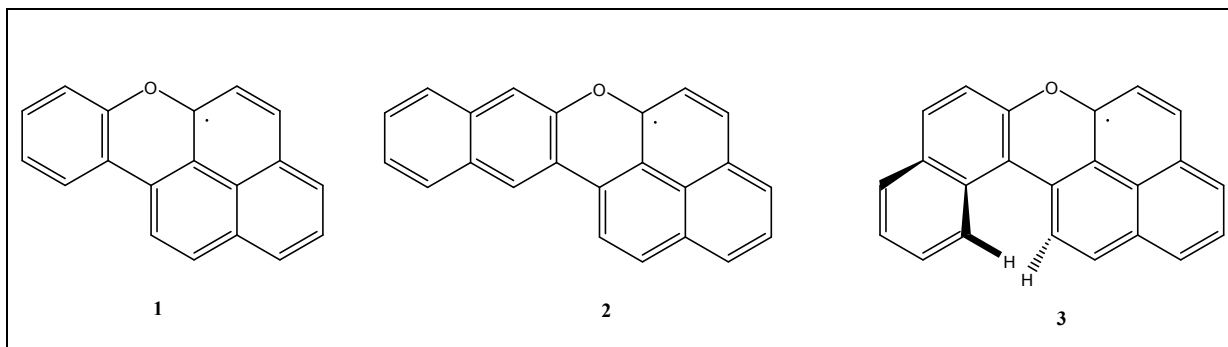
Stable Phenalenyl Radicals Showing Reversible Redox Chemistry Under Air

Götz Bucher,^a Ommid Anamimoghadam,^a Santi Nonell,^b
and Lee Cronin^a

^a WestCHEM, School of Chemistry, University of Glasgow, University Avenue,
Glasgow G12 8QQ
(goetz.bucher@gla.ac.uk)

^b Institut Químic de Sarrià, Universitat Ramon Llull, Barcelona, Spain

Stable free radicals based on the phenalenyl moiety for some time have been a chromophore of choice, if an organic framework capable of existing in different redox states was required, with nearly invariant bond orders.^{1,2} For stability towards dimerization, however, phenalenyl radicals normally would require steric protection, hindering spin-spin communication. In this contribution, we present new data on a series of phenalenyl-based free radicals such as **1-3**, which are stabilized electronically via a ring oxygen atom, and do not require steric protection.³ We will outline the synthesis, characterization, and properties of **1-3**, and present data that indicate that these radicals show unusual stability, both thermal, and towards oxygen.



1. Chi X.; Itkis M.E.; Patrick B.O.; Barclay T.M.; Reed R.W.; Oakley R.T.; Cordes A.W.; Haddon R.C. *J. Am. Chem. Soc.* **1999**, *121*, 10395.

2. Liao P.; Itkis M.E.; Oakley R.T.; Tham F.S.; Haddon R.C. *J. Am. Chem. Soc.* **2004**, *126*, 14297.

3. Anamimoghadam, O.; Symes, M.D.; Busche, C.; Long, D.-L.; Caldwell, S.T.; Flors, C.; Nonell, S.; Cronin, L.; Bucher, G. *Org. Lett.* **2013**, *15*, 2970.